

ToppersNotes

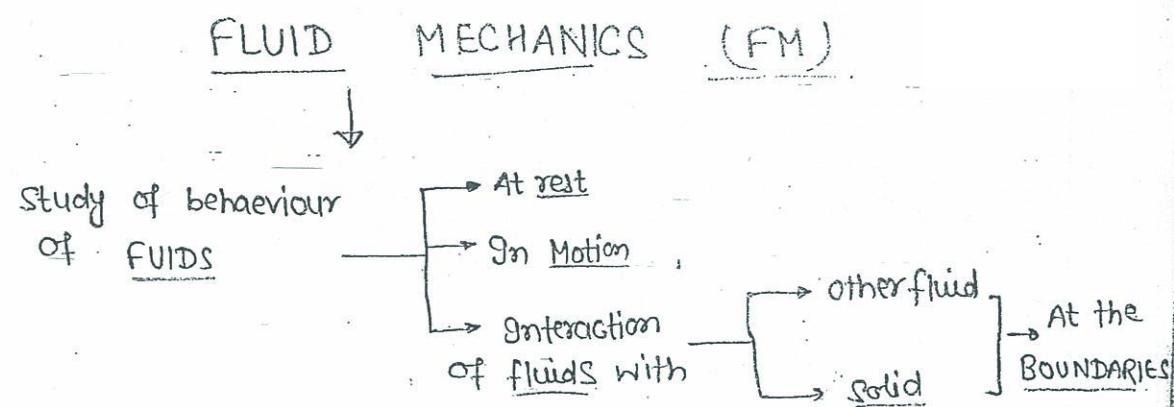
**IES/GATE
MECHANICAL ENGINEERING**

**FLUID MECHANICS
&
RAC, REFRIGERATION, AIR CONDITIONING**

VOLUME-I

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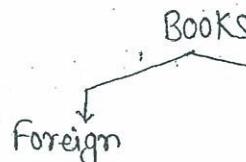
Syllabus :- FM

- ① Introduction — Fluids and Properties.
- ② Pressure & its Measurement.
- ③ Hydrostatic forces on surfaces.
- ④ Buoyancy and floatation.
- ⑤ Fluid Kinematics.
- ⑥ Fluid Dynamics → Motion: Greek word.
- ⑦ flow through pipes.
- ⑧ Laminar flow.
- ⑨ Turbulent flow through pipes.
- ⑩ Boundary layer Theory.
- ⑪ Dimensional Analysis.

Study Material

① Class Notes

② Books



- Fox & McDonald
- Cengell
- Shames

- R.K Bansal

- Modiseth

- Kumar

- Subramaniam (Que).

MANOJ NAUDIYAL

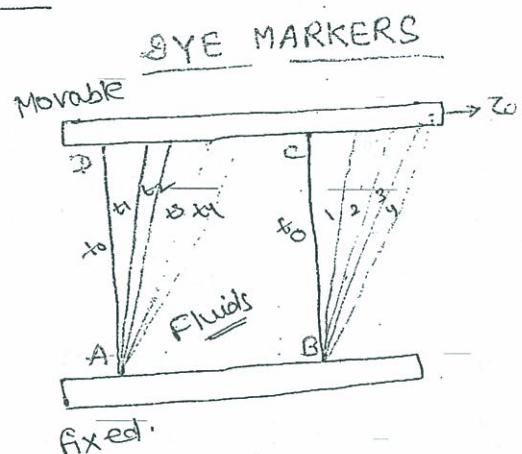
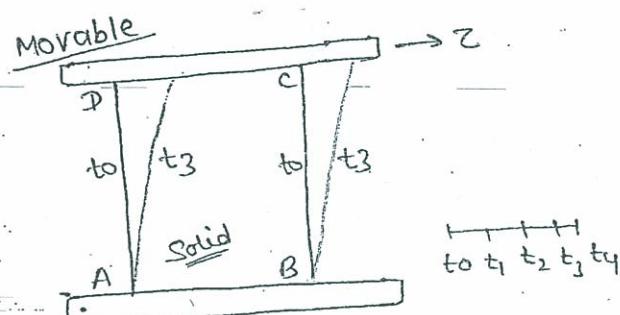
CHAPTER - 1 :-INTRODUCTIONFLUIDS & PROPERTIES

Behaviour under forces / stresses

Solid fluids.

- ① Tensile ✓
- ② Compressive ✓
- ③ shear ✓

Behaviour Under shear stress



- ① No further deformation
(after t_3 upto elastic limit
not crossed)
- ② Regains original shape.

Concept of fluid :-

fluid is defined as a substance which can't sustain shear stress at rest.

Fluid can also be defined as substance which deforms continuously under action of a shear stress, no matter how small the shear stress may be.

Fluid is a substance which is capable of flowing and changing its shape, this is because of the

its inability to resist shearing stresses.

However this does not mean that the fluids do not offer any resistance to the shearing forces.

In fact as the fluids flow, there exist tangential or shearing stresses between the layers of the fluid due to viscosity. However as long as the fluid is at rest, no shear stress can exist within it.

Ideal Fluid :-

It is an imaginary and hypothetical fluid which has

- i) Zero viscosity.
- ii) Zero surface tension.
- iii) It is perfectly incompressible

Because of zero viscosity, the ideal fluid encounters zero resistance to its motion.

Concept of Continuum :-

Continuum implies a continuous distribution of matter with no voids or empty spaces. For mechanical analysis fluids are assumed as continuum because even the gases have a very high molecular density and the distances between them the molecules is very very small.

Concept of continuum breaks down in

- i) High vacuum condition.
- ii) Rarefied gas flows
- iii) Shock

No slip Condition :-

Fluid in direct contact with the solid boundary has the velocity of the boundary itself, in other words the layer of the fluid sticks to the solid boundary and "there is no slip between the fluid and the solid at the contact".

PROPERTIES OF FLUIDS:-(1) Mass Density or specific mass (ρ)

$$\rho = \frac{\text{mass}}{\text{volume}} \quad \text{or} \quad \rho = \frac{m}{V} \quad \text{kg/m}^3.$$

$$P \uparrow \Rightarrow \rho \uparrow$$

$$T \uparrow \Rightarrow \rho \downarrow$$

(2) Weight density or specific weight (w or γ)

$$w = \frac{\text{Wt.}}{\text{Volume}} = \frac{N}{m^3}$$

$$w = \frac{mg}{V} = \frac{Vg}{V} \quad \text{or} \quad w = \rho g$$

(3) Specific volume (V)

$$V = \frac{\text{Vol}}{\text{mass}} \quad \text{m}^3/\text{kg}$$

$$V = \frac{1}{\rho}$$

(4) Specific gravity
or
Relative density

$$S = \frac{P_{\text{test fluid}}}{P_{\text{std. fluid}}} = \frac{w_{\text{test fluid}}}{w_{\text{std. fluid}}}$$

Standard fluid :- liquid — water at 4°C .

Liquid - water at 4°C

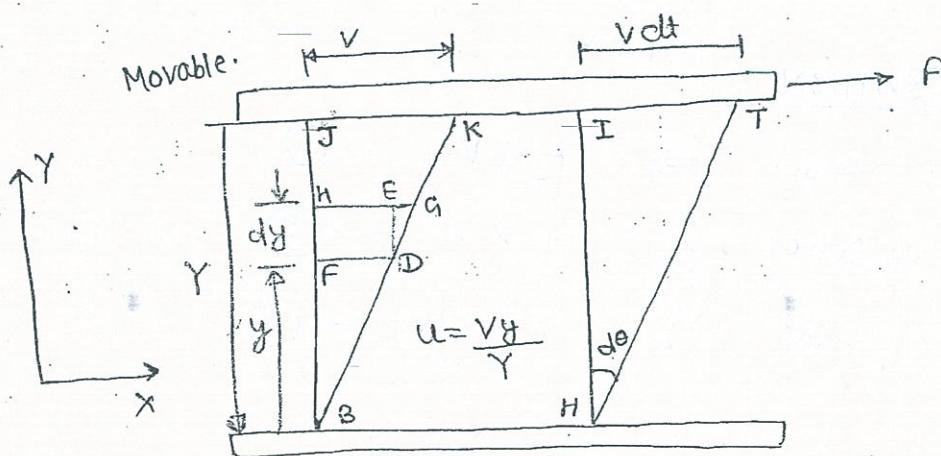
Gas - Air / N_2 at specified Temp & Pres.

(S) Viscosity

It is the property by virtue of which fluid offers resistance to the movement of one layer of fluid over adjacent layers.

It is due to the forces of cohesion and molecular momentum exchange between the layers of the fluid and as the flow takes place this effect appears as shearing stresses between the layers of the fluids.

Newton's Law of Viscosity :-

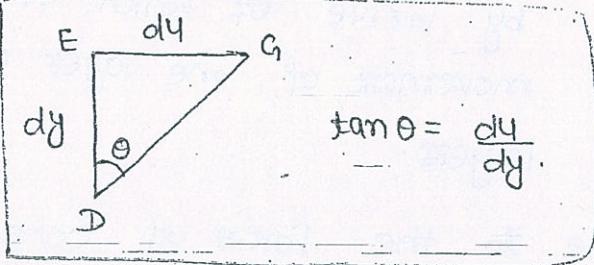


$$ED = HF = u$$

$$FH = \omega E = dy \quad EG = du$$

Assumption:-

- ① F = shear force on the upper plate.
- ② A = Area of the upper plate in contact with the fluid.
- ③ V = Velocity of the upper plate.
- ④ γ = Distance b/w the plates.
- ⑤ V and γ are not very large (i.e. the variation to be linear).

IMP.In Similar $\triangle BJK$ & BFD .

$$\frac{JK}{JB} = \frac{FD}{FB} \quad \text{or} \quad \frac{\gamma}{r} = \frac{U}{y}$$

or $U = \frac{VY}{Y}$ \Rightarrow linear velocity profile.

F A V Y.

OBSERVATION

- (i) $A, V = \text{const.} \Rightarrow F \propto V \rightarrow ①$
- (ii) $\gamma, V = \text{const.} ; F \propto A \rightarrow ②$
- (iii) $A \& V = \text{const.} ; F \propto \frac{1}{\gamma} \rightarrow ③$

Combining ① ② & ③

$$F \propto \frac{AV}{r} \quad \text{or} \quad \frac{F}{A} \propto \frac{V}{r}$$

$$\tau \propto \frac{V}{r} \quad \text{--- } ④$$

In similar triangle BJK and DEG ;

$$\frac{JK}{JB} = \frac{EG}{ED}$$

$$\frac{V}{r} = \frac{du}{dy} \quad \text{--- } ⑤$$

velocity gradient

Using ⑤ in. ④ ;

$$\tau \propto \frac{du}{dy}$$

IMP

$$\tau = \mu \frac{du}{dy} \quad \text{--- } ⑥$$

Newton's law.

μ = Constant of Proportionality.

= Dynamic / Absolute viscosity.

UNITS :- SI UNIT

* $\frac{\text{N-S}}{\text{m}^2}$ or Pa.sec = kg/m-sec

* 1 Poise = 1 Pa.sec.

μ It is the shear stress required to produce unit velocity gradient or deformation rate.

Kinematic viscosity (v)

$$\nu = \frac{\mu}{\rho}$$

m^2/sec

Cgs Unit

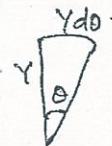
$$1 \text{ Stoke} = 1 \text{ cm}^2/\text{sec.}$$

In ΔHIT :

$$\Delta T = v dt = \gamma \cdot d\theta$$

$$\frac{d\theta}{dt} = \frac{v}{\gamma} \quad \text{--- (7)}$$

Using (5) in (7)



IMP.

$$\frac{d\theta}{dt} = \frac{du}{dy} \quad \text{--- (8)}$$

$$\Rightarrow \frac{\text{deformation rate}}{\text{rate}} = \frac{\text{velocity gradient}}{\text{gradient}}$$

Shear stress is always associated with deformation rate.

Effect of Temp. \uparrow on viscosity :-

	u	v
Liquid	↓	↓
gases	↑	↑

* Classification of fluids on the basis of relation between τ and $\frac{du}{dy}$.

Fluids

Newtonian

— obeys Newton's law. $\left[\frac{dy}{dp} \propto z \right]$

= viscosity = $f(\text{temp})$

→ independent of $\frac{du}{dy}$, time

Ex:- Water, air, Petrol, diesel, Kerosene, Mercury, all gases, glycerine, ethyl alcohol, Methyl alcohol etc.

Non-Newtonian fluids

→ obeys Power law:

$$z = z_y + K \cdot \left(\frac{du}{dy} \right)^n \quad (1)$$

z_y = shear yield stress

K = consistency index

n = flow behaviour index

eqn (1) can be written as;

$$z = z_y + \left\{ K \cdot \left(\frac{du}{dy} \right)^{n-1} \right\} \left(\frac{du}{dy} \right)$$

↓
 η

η = Apparent viscosity.

$$z = z_y + \eta \frac{du}{dy}$$

Non-Newtonian fluids :-i) Time Independent :-

$$z_y = 0 ; \text{ viscosity} = f(\text{Temp}, \frac{du}{dy})$$

→ independent of time.

ii) Time Dependent.

\Rightarrow Time Independent

Pseudoplastic

or Shear thinning.

$$\Rightarrow \tau_y = 0, n < 1$$

$$\text{As. } \frac{du}{dy} \uparrow \Rightarrow \text{visco.} \downarrow$$

Ex:- Paint, Blood, Milk,
Polymer solution, Colloids,
paper pulp in water,
Ketchup etc.

Dilatant or shear

Thickening :-

$$\tau_y = 0, n > 1$$

$$\text{As. } \frac{du}{dy} \uparrow \Rightarrow \text{visco.} \uparrow$$

Ex:- Butter solution,

Sugar syrup,

Corn flour,

Quick sand (जलमणि)

(ii) Time Dependent Non-Newtonian Fluids

$$\tau_y > 0; \text{ viscosity} = f(\text{temp.}, \frac{du}{dy}, \text{time})$$

Thixotropic

$$\tau_y > 0, n < 1$$

$$\text{As time} \uparrow = \text{visc.} \downarrow$$

Ex:- Ice cream.

YOGHURT

MARGARINIC

PRINTER INK etc.

Rheopectic

$$\tau_y > 0, n > 1$$

$$\text{As time} \uparrow \Rightarrow \text{visc.} \uparrow$$

Ex:- Gypsum paste.

Bentonite solution
etc.

Rheology

\hookrightarrow study of Non. Newtonian fluids.

Bingham or Ideal Plastic :-

Ex:- Tooth paste

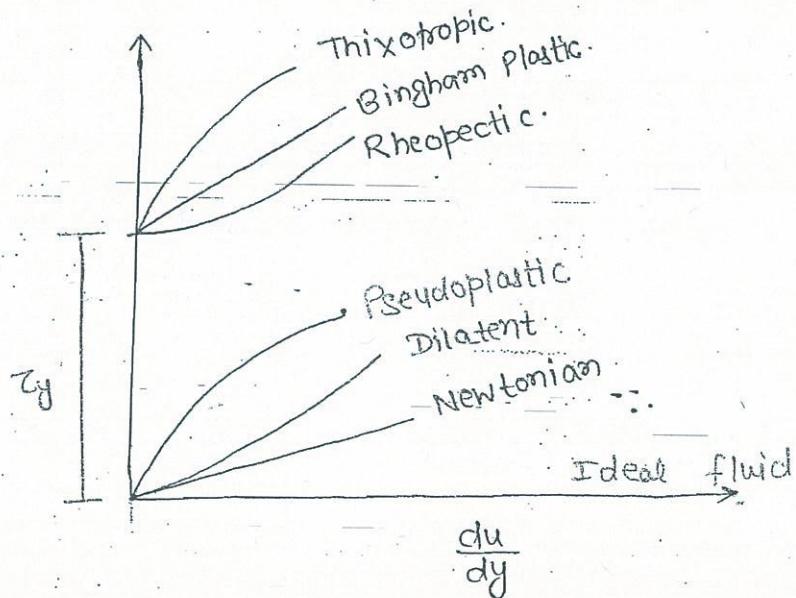
Drilling mud.

$$\text{Power law; } \tau = \tau_y + K \cdot \left(\frac{du}{dy} \right)^n$$

for Bingham Plastic.

$$\tau_y > 0; \quad K = \mu, \quad n = 1$$

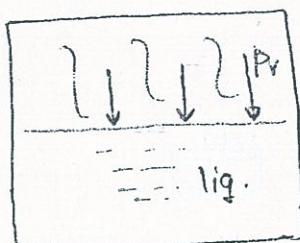
$$\therefore \boxed{\tau = \tau_y + \mu \cdot \left(\frac{du}{dy} \right)} \quad - \text{Power law for Bingham plastic.}$$



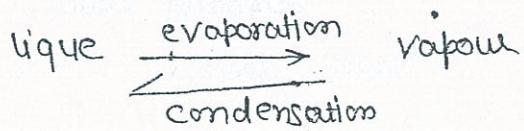
Bingham plastic is a fluid which behaves like a solid - until τ_y is exceeded and after that it behaves like a Newtonian fluid.

Drilling mud is used to displace the mud from the oil well during extraction.

6) Vapour Pressure :-



Phase equilibrium.



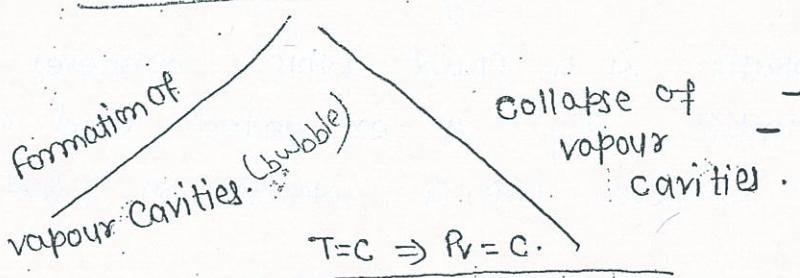
All liquids have a tendency to vapourise. Vapour pressure is defined as the pressure exerted by vapour of a liquid which are in phase equilibrium with the liquid itself.

Vapour pressure increases with temperature.

If the External absolute pressure exerted on the liquid and the vapour pressure of the liquid become equal, then the phenomenon of Boiling starts taking place.

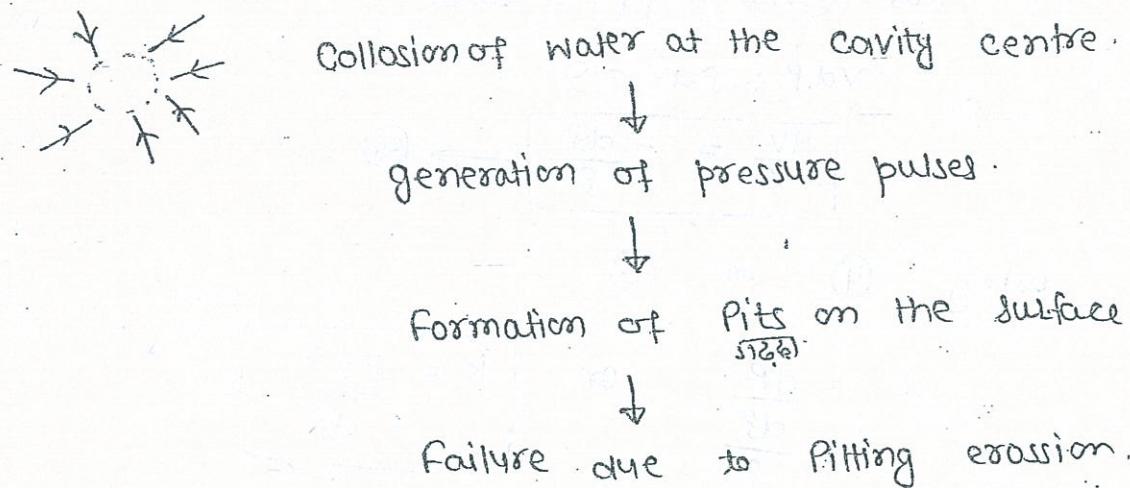
Boiling is not spontaneous but evaporation is even at 0°C .

CAVITATION:-



$$\begin{array}{c}
 + A \\
 \dots \\
 P_A < P_v \\
 45 < 50
 \end{array}
 \quad
 \begin{array}{c}
 + B \\
 \dots \\
 P_B > P_v \\
 52 > 50
 \end{array}$$

P_v can be more is achieved by the arrangement of tube.



Methods to prevent cavitation

Outlet of Reaction turbine Inlet of centrifugal pumps.

Draft tube is provided.



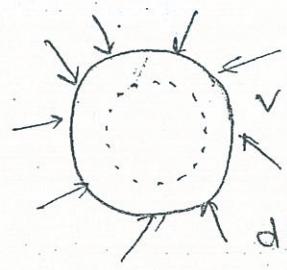
NPSH
 (Net Pump suction head)

(7) Compressibility and Elasticity :-

(K)

↳ Bulk modulus of elasticity.

$$\beta = \frac{1}{K} \quad \textcircled{1}$$



$$K = \frac{dP}{-\frac{\partial V}{V}} \quad \textcircled{2}$$

$$K = -\frac{V dP}{\partial V} \quad \textcircled{3}$$

$$\text{mass } (m) = \rho V = \text{const.}$$

$$V dP + P dV = 0$$

$$\frac{dV}{V} = -\frac{dP}{P} \rightarrow \textcircled{4}$$

using \textcircled{4} in \textcircled{3}

$$K = \frac{dP}{d\frac{\rho}{P}}$$

$$K = \frac{\rho dP}{dP}$$

$$\beta = \frac{1}{K} \quad \text{or} \quad \beta = \frac{dP}{\rho dP} \rightarrow \textcircled{5}$$

if $dP = 0 \Rightarrow \beta = 0 \Rightarrow \text{Incompressible fluid.}$

$$K = 0.103 \text{ MPa} \quad (\text{Air})$$

$$K = 2060 \text{ MPa} \quad (\text{H}_2\text{O})$$

$$K = 2060,00 \text{ MPa} \quad (\text{Steel})$$

for water the value of β is very less so under normal condition the water is Incompressible.

C_p = Compressibility factor.

$$= 1 + \frac{M^2}{4} + \frac{2-\gamma}{24} M^4 + \dots$$

where: γ = Ratio of specific heats

$$M = \text{Mach. No} = \frac{V}{c} = \frac{\text{velocity of air}}{\text{velocity of sound in air}}$$

$$\text{if } \gamma = 1.4.$$

$$C_p = 1 + \frac{M^2}{4} + \frac{M^4}{40} + \dots$$

$$M \quad \left. \begin{array}{l} \% \text{ error} \\ \% \text{ change in density} \end{array} \right\} = (C_p - 1) \times 100$$

0.1	25
0.2	50
0.3	75
0.4	100
1.0	27.5

IMP.

$$M < 0.2 - 0.3$$

when the Mach no lies in b/w 0.2 - 0.3 then the % age change in density is less than 5%.

IMP: Generally the liquids are taken as Incompressible because their value of K is very high.

Even the gases can be taken as Incompressible if the value of Mach No "M" is less than $M < 0.2 - 0.3$ because for this range the percentage change in density is much less than 5%.

G.8. W.B.

$$K = \frac{\rho \cdot dp}{dp}$$

$$P = (3500 \rho^{1/2} + 2500)$$

$$K = \rho \cdot \frac{1750}{\rho^{1/2}}$$

$$\frac{dp}{dp} = \frac{3500 \times 1750}{\rho^{1/2}}$$

$$= \rho^{-1 - \frac{1}{2}} \cdot 1750$$

$$= \frac{1750}{\rho^{1/2}}$$

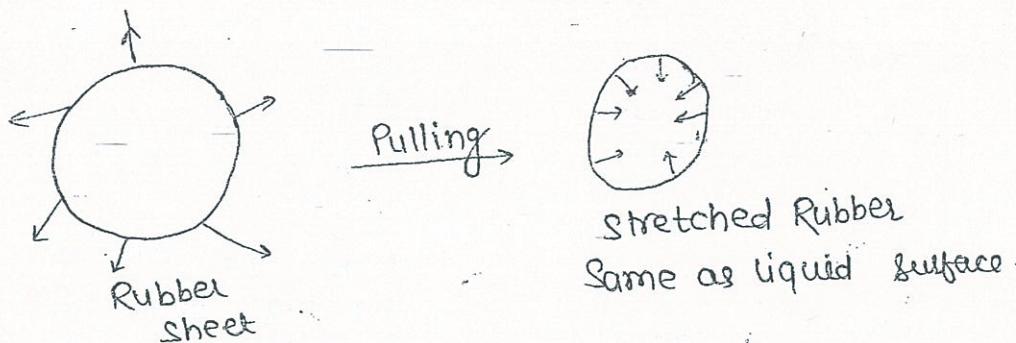
$$= (776)^{1/2} \times 1750$$

$$(100 \text{ kPa}) = (3500 \rho^{1/2} + 2500) \frac{N}{m^2}$$

$$= 48750 \frac{N}{m^2}$$

$$\rho = 776$$

* SURFACE TENSION :- (σ)



Definition :-

It is the property by virtue of which a liquid tends to minimize its own surface area.

Concept of Surface Tension :-

Drops of liquid behave like small spherical balloons filled with liquid and the surface of the drops acts like stretched elastic membrane under tension. The pulling force which causes this tension is due to the force of cohesion and acts parallel to the surface. It is the intensity of these forces per unit length of the surface that gives the value of surface tension.

Surface Energy :-

Surface Tension can also be defined as surface energy. In this case it represents the amount of work required to be done in order to increase the area of the surface by unit amount.

σ always act along the length.